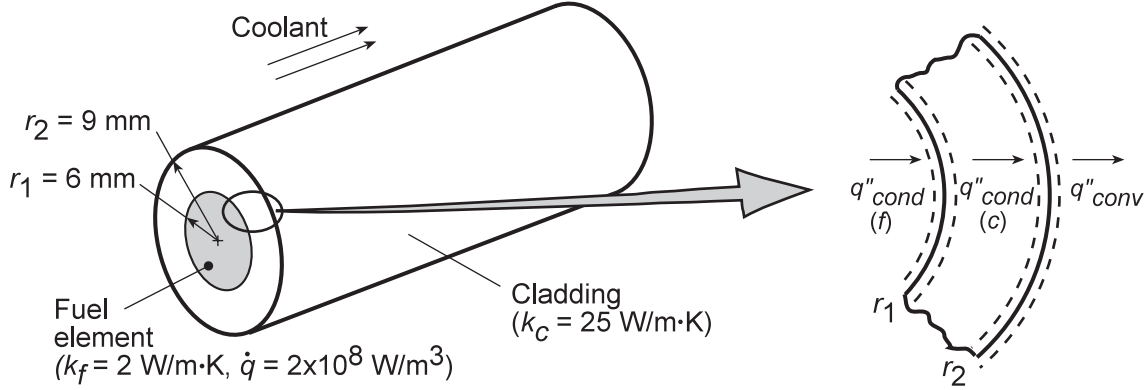


PROBLEM 3.98

KNOWN: Radii and thermal conductivities of reactor fuel element and cladding. Fuel heat generation rate. Temperature and convection coefficient of coolant.

FIND: (a) Expressions for temperature distributions in fuel and cladding, (b) Maximum fuel element temperature for prescribed conditions, (c) Effect of h on temperature distribution.

SCHEMATIC:



ASSUMPTIONS: (1) Steady-state conditions, (2) One-dimensional conduction, (3) Negligible contact resistance, (4) Constant properties.

ANALYSIS: (a) From Eqs. 3.54 and 3.28, the heat equations for the fuel (f) and cladding (c) are

$$\frac{1}{r} \frac{d}{dr} \left(r \frac{dT_f}{dr} \right) = -\frac{\dot{q}}{k_f} \quad (0 \leq r \leq r_1) \quad \frac{1}{r} \frac{d}{dr} \left(r \frac{dT_c}{dr} \right) = 0 \quad (r_1 \leq r \leq r_2)$$

Hence, integrating both equations twice,

$$\frac{dT_f}{dr} = -\frac{\dot{q}r}{2k_f} + \frac{C_1}{k_f r} \quad T_f = -\frac{\dot{q}r^2}{4k_f} + \frac{C_1}{k_f} \ln r + C_2 \quad (1,2)$$

$$\frac{dT_c}{dr} = \frac{C_3}{k_c} \quad T_c = \frac{C_3}{k_c} \ln r + C_4 \quad (3,4)$$

The corresponding boundary conditions are:

$$\left. \frac{dT_f}{dr} \right|_{r=0} = 0 \quad T_f(r_1) = T_c(r_1) \quad (5,6)$$

$$\left. -k_f \frac{dT_f}{dr} \right|_{r=r_1} = \left. -k_c \frac{dT_c}{dr} \right|_{r=r_1} \quad \left. -k_c \frac{dT_c}{dr} \right|_{r=r_2} = h [T_c(r_2) - T_\infty] \quad (7,8)$$

Note that Eqs. (7) and (8) are obtained from surface energy balances at r_1 and r_2 , respectively. Applying Eq. (5) to Eq. (1), it follows that $C_1 = 0$. Hence,

$$T_f = -\frac{\dot{q}r^2}{4k_f} + C_2 \quad (9)$$

From Eq. (6), it follows that

$$-\frac{\dot{q}r_1^2}{4k_f} + C_2 = \frac{C_3 \ln r_1}{k_c} + C_4 \quad (10)$$

Continued...

PROBLEM 3.98 (Cont.)

Also, from Eq. (7),

$$\frac{\dot{q}r_1}{2} = -\frac{C_3}{r_1} \quad \text{or} \quad C_3 = -\frac{\dot{q}r_1^2}{2} \quad (11)$$

Finally, from Eq. (8), $-\frac{C_3}{r_2} = h \left[\frac{C_3}{k_c} \ln r_2 + C_4 - T_\infty \right]$ or, substituting for C_3 and solving for C_4

$$C_4 = \frac{\dot{q}r_1^2}{2r_2h} + \frac{\dot{q}r_1^2}{2k_c} \ln r_2 + T_\infty \quad (12)$$

Substituting Eqs. (11) and (12) into (10), it follows that

$$C_2 = \frac{\dot{q}r_1^2}{4k_f} - \frac{\dot{q}r_1^2}{2k_c} \ln r_1 + \frac{\dot{q}r_1^2}{2r_2h} + \frac{\dot{q}r_1^2}{2k_c} \ln r_2 + T_\infty$$

$$C_2 = \frac{\dot{q}r_1^2}{4k_f} + \frac{\dot{q}r_1^2}{2k_c} \ln \frac{r_2}{r_1} + \frac{\dot{q}r_1^2}{2r_2h} + T_\infty \quad (13)$$

Substituting Eq. (13) into (9),

$$T_f = \frac{\dot{q}}{4k_f} (r_1^2 - r^2) + \frac{\dot{q}r_1^2}{2k_c} \ln \frac{r_2}{r_1} + \frac{\dot{q}r_1^2}{2r_2h} + T_\infty \quad (14)$$

Substituting Eqs. (11) and (12) into (4),

$$T_c = \frac{\dot{q}r_1^2}{2k_c} \ln \frac{r_2}{r} + \frac{\dot{q}r_1^2}{2r_2h} + T_\infty \quad (15)$$

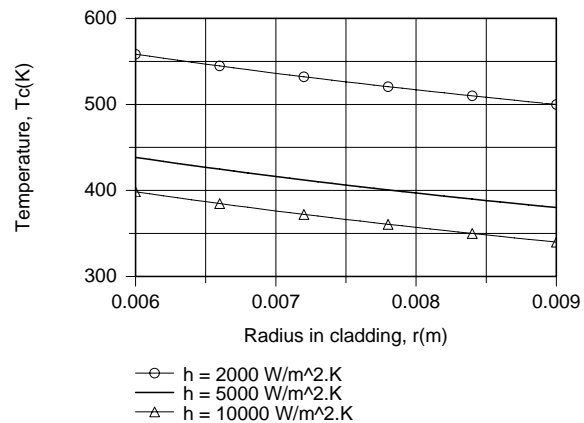
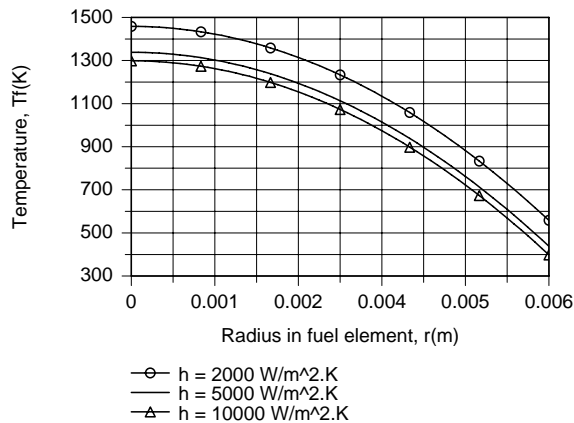
(b) Applying Eq. (14) at $r = 0$, the maximum fuel temperature for $h = 2000 \text{ W/m}^2 \cdot \text{K}$ is

$$T_f(0) = \frac{2 \times 10^8 \text{ W/m}^3 \times (0.006 \text{ m})^2}{4 \times 2 \text{ W/m} \cdot \text{K}} + \frac{2 \times 10^8 \text{ W/m}^3 \times (0.006 \text{ m})^2}{2 \times 25 \text{ W/m} \cdot \text{K}} \ln \frac{0.009 \text{ m}}{0.006 \text{ m}}$$

$$+ \frac{2 \times 10^8 \text{ W/m}^3 (0.006 \text{ m})^2}{2 \times (0.009 \text{ m}) 2000 \text{ W/m}^2 \cdot \text{K}} + 300 \text{ K}$$

$$T_f(0) = (900 + 58.4 + 200 + 300) \text{ K} = 1458 \text{ K}.$$

(c) Temperature distributions for the prescribed values of h are as follows:



Continued...

PROBLEM 3.98 (Cont.)

Clearly, the ability to control the maximum fuel temperature by increasing h is limited, and even for $h \rightarrow \infty$, $T_f(0)$ exceeds 1000 K. The overall temperature drop, $T_f(0) - T_\infty$, is influenced principally by the low thermal conductivity of the fuel material.

COMMENTS: For the prescribed conditions, Eq. (14) yields, $T_f(0) - T_f(r_1) = \dot{q} r_1^2 / 4k_f = (2 \times 10^8 \text{ W/m}^3)(0.006 \text{ m})^2 / 8 \text{ W/m}\cdot\text{K} = 900 \text{ K}$, in which case, with no cladding and $h \rightarrow \infty$, $T_f(0) = 1200 \text{ K}$. To reduce $T_f(0)$ below 1000 K for the prescribed material, it is necessary to reduce \dot{q} .