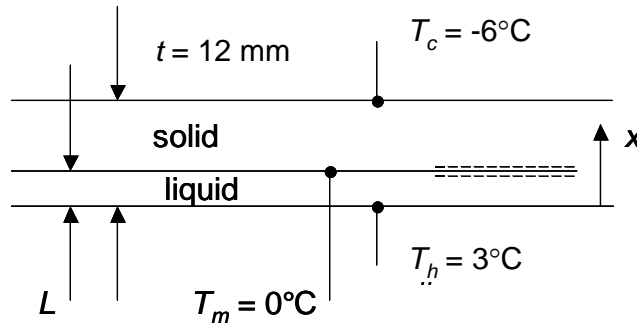


PROBLEM 3.8

KNOWN: Top and bottom temperatures applied to a water layer of known thickness.

FIND: Steady-state location of the solid-liquid interface.

SCHEMATIC:



ASSUMPTIONS: (1) Steady-state conditions, (2) Constant properties, (3) Negligible radiation, (4) Negligible convection in the liquid.

PROPERTIES: Table A.6, liquid water ($T = 273 \text{ K}$): $k_f = 0.569 \text{ W/m}\cdot\text{K}$; Table A.3, ice ($T = 0 \text{ K}$), $k_s = 1.88 \text{ W/m}\cdot\text{K}$.

ANALYSIS: An energy balance at the control surface shown in the schematic yields

$$k_f(T_h - T_m)/L = k_s(T_m - T_c)/(t - L)$$

or

$$L = \frac{t}{\left(\frac{k_s(T_m - T_c)}{k_f(T_h - T_m)} + 1 \right)} = \frac{12 \times 10^{-3} \text{ m}}{\left(\frac{1.88 \text{ W/m}\cdot\text{K}(0^\circ\text{C} - (-6^\circ\text{C}))}{0.569 \text{ W/m}\cdot\text{K}(3^\circ\text{C} - 0^\circ\text{C})} + 1 \right)} = 1.58 \times 10^{-3} \text{ m} = 1.58 \text{ mm} \quad <$$

COMMENTS: (1) Liquid water is opaque to thermal radiation, but ice is semi-transparent. A more detailed analysis would account for the effects of radiation. (2) Free convection in the liquid is negligible because the density of liquid water at $T_h = 3^\circ\text{C}$ is greater than the density at $T_m = 0^\circ\text{C}$. Water is one of only several liquids that experiences such a *density inversion*.