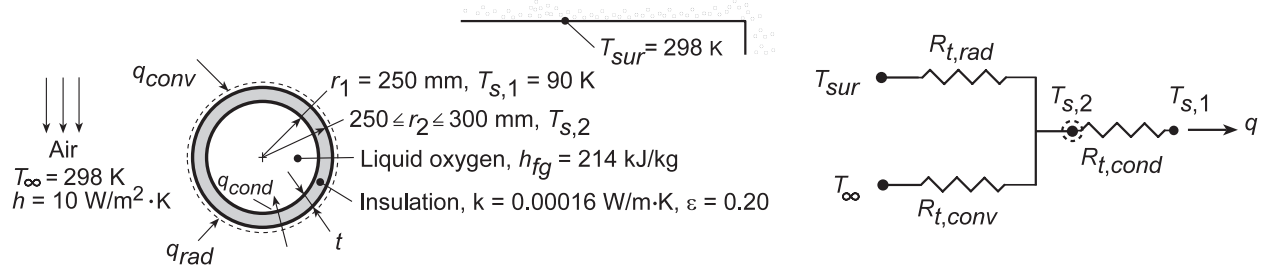


### PROBLEM 3.64

**KNOWN:** Diameter of a spherical container used to store liquid oxygen and properties of insulating material. Environmental conditions.

**FIND:** (a) Reduction in evaporative oxygen loss associated with a prescribed insulation thickness, (b) Effect of insulation thickness on evaporation rate.

**SCHEMATIC:**



**ASSUMPTIONS:** (1) Steady-state, one-dimensional conduction, (2) Negligible conduction resistance of container wall and contact resistance between wall and insulation, (3) Container wall at boiling point of liquid oxygen.

**ANALYSIS:** (a) Applying an energy balance to a control surface about the insulation,  $\dot{E}_{in} - \dot{E}_{out} = 0$ , it follows that  $q_{conv} + q_{rad} = q_{cond} = q$ . Hence,

$$\frac{T_{\infty} - T_{s,2}}{R_{t,conv}} + \frac{T_{sur} - T_{s,2}}{R_{t,rad}} = \frac{T_{s,2} - T_{s,1}}{R_{t,cond}} = q \quad (1)$$

where  $R_{t,conv} = (4\pi r_2^2 h)^{-1}$ ,  $R_{t,rad} = (4\pi r_2^2 h_r)^{-1}$ ,  $R_{t,cond} = (1/4\pi k)[(1/r_1) - (1/r_2)]$ , and, from Eq.

1.9, the radiation coefficient is  $h_r = \varepsilon \sigma (T_{s,2} + T_{sur}) (T_{s,2}^2 + T_{sur}^2)$ . With  $t = 10$  mm ( $r_2 = 260$  mm),  $\varepsilon =$

0.2 and  $T_{\infty} = T_{sur} = 298$  K, an iterative solution of the energy balance equation yields  $T_{s,2} \approx 297.7$  K, where  $R_{t,conv} = 0.118$  K/W,  $R_{t,rad} = 0.982$  K/W and  $R_{t,cond} = 76.5$  K/W. With the insulation, it follows that the heat gain is

$$q_w \approx 2.72 \text{ W}$$

Without the insulation, the heat gain is

$$q_{wo} = \frac{T_{\infty} - T_{s,1}}{R_{t,conv}} + \frac{T_{sur} - T_{s,1}}{R_{t,rad}}$$

where, with  $r_2 = r_1$ ,  $T_{s,1} = 90$  K,  $R_{t,conv} = 0.127$  K/W and  $R_{t,rad} = 3.14$  K/W. Hence,

$$q_{wo} = 1702 \text{ W}$$

With the oxygen mass evaporation rate given by  $\dot{m} = q/h_{fg}$ , the percent reduction in evaporated oxygen is

$$\% \text{ Reduction} = \frac{\dot{m}_{wo} - \dot{m}_w}{\dot{m}_{wo}} \times 100\% = \frac{q_{wo} - q_w}{q_{wo}} \times 100\%$$

Hence,

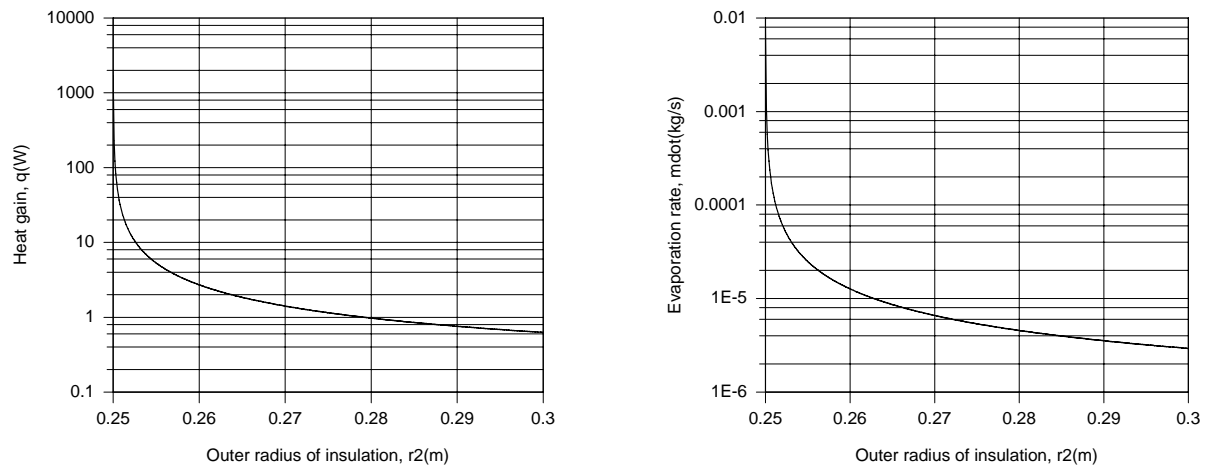
$$\% \text{ Reduction} = \frac{(1702 - 2.7) \text{ W}}{1702 \text{ W}} \times 100\% = 99.8\%$$

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Continued...

### PROBLEM 3.64 (Cont.)

(b) Using Equation (1) to compute  $T_{s,2}$  and  $q$  as a function of  $r_2$ , the corresponding evaporation rate,  $\dot{m} = q/h_{fg}$ , may be determined. Variations of  $q$  and  $\dot{m}$  with  $r_2$  are plotted as follows.



Because of its extremely low thermal conductivity, significant benefits are associated with using even a thin layer of insulation. Nearly three-order magnitude reductions in  $q$  and  $\dot{m}$  are achieved with  $r_2 = 0.26$  m. With increasing  $r_2$ ,  $q$  and  $\dot{m}$  decrease from values of 1702 W and  $8 \times 10^{-3}$  kg/s at  $r_2 = 0.25$  m to 0.627 W and  $2.9 \times 10^{-6}$  kg/s at  $r_2 = 0.30$  m.

**COMMENTS:** Laminated metallic-foil/glass-mat insulations are extremely effective and corresponding conduction resistances are typically much larger than those normally associated with surface convection and radiation.